

Phosphors Processing and their Properties

the case of SrS:Ce

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Regular Luminescence of SrS:Ce³⁺

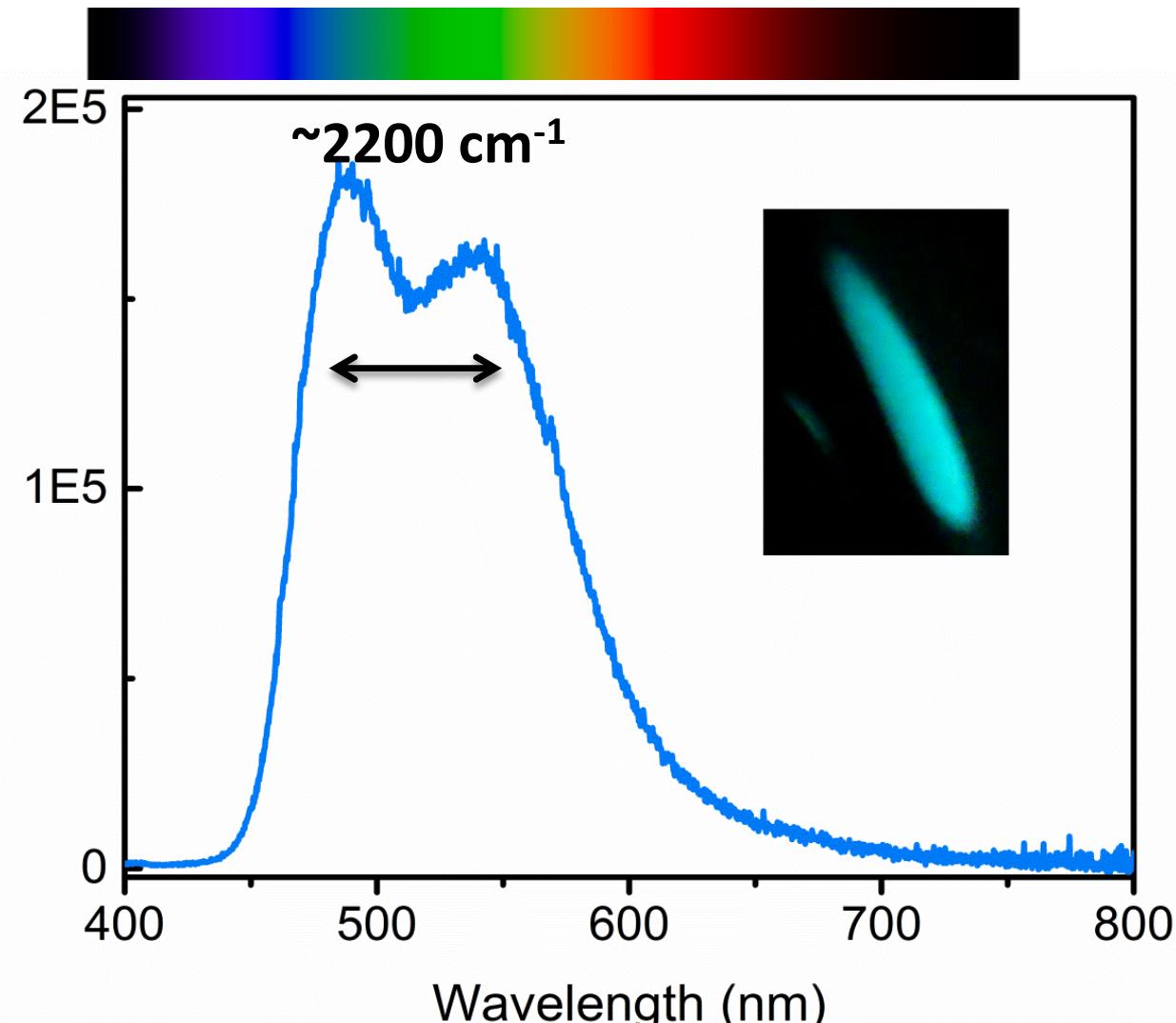
Ce³⁺ - 4f¹



5d

2F_{7/2}

2F_{5/2}



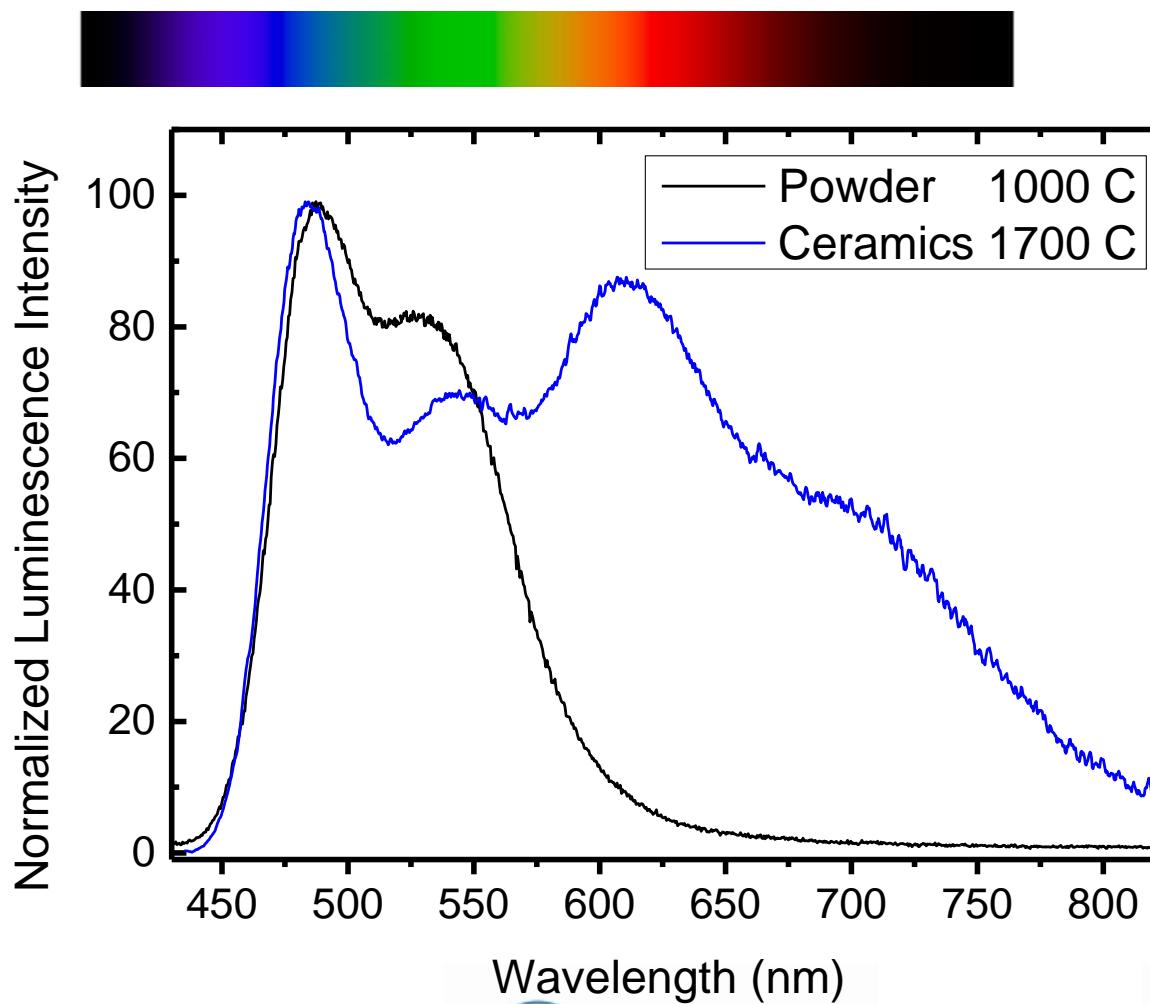
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A New PL of SrS:Ce³⁺ Ceramics in Red (+IR)



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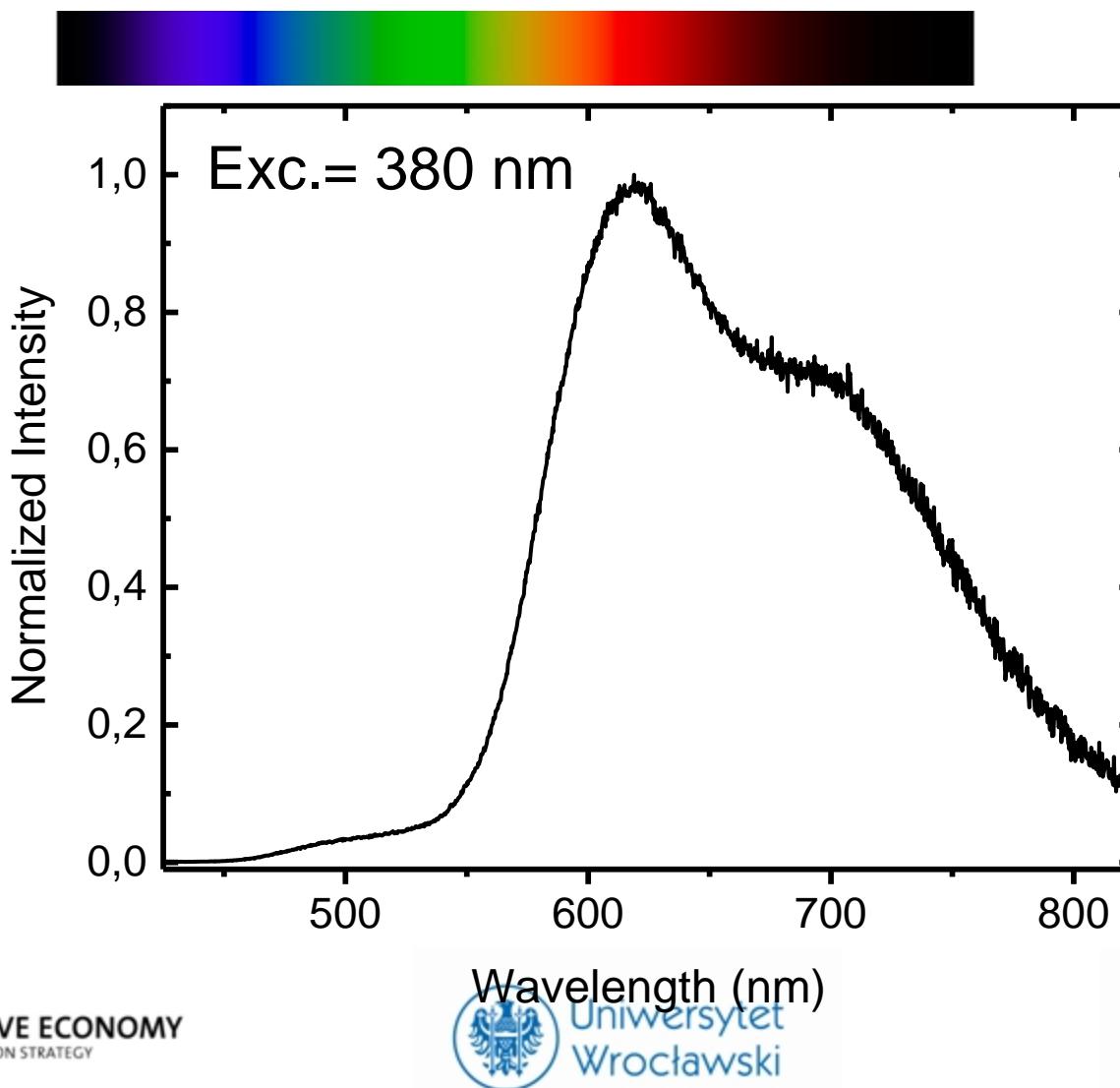


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The Red (+IR) PL may be excited selectively



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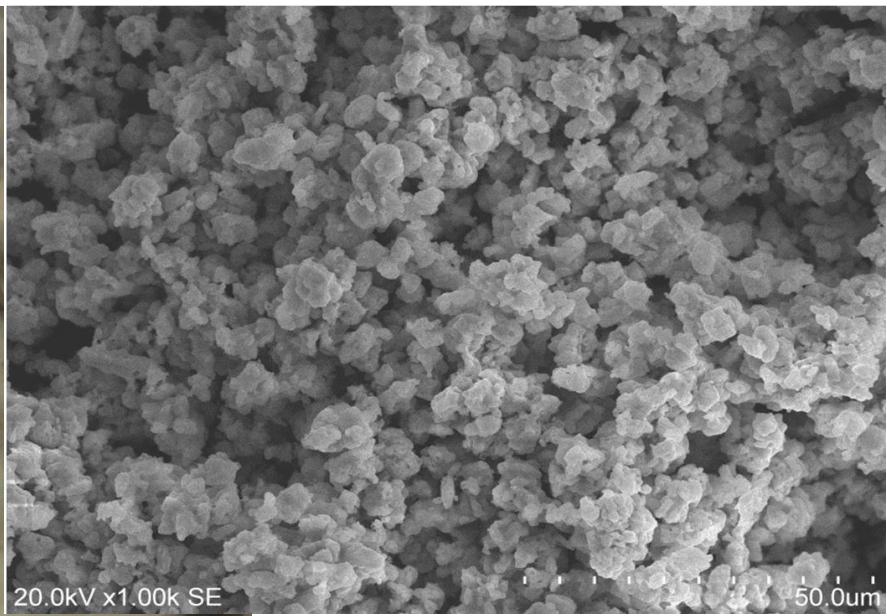
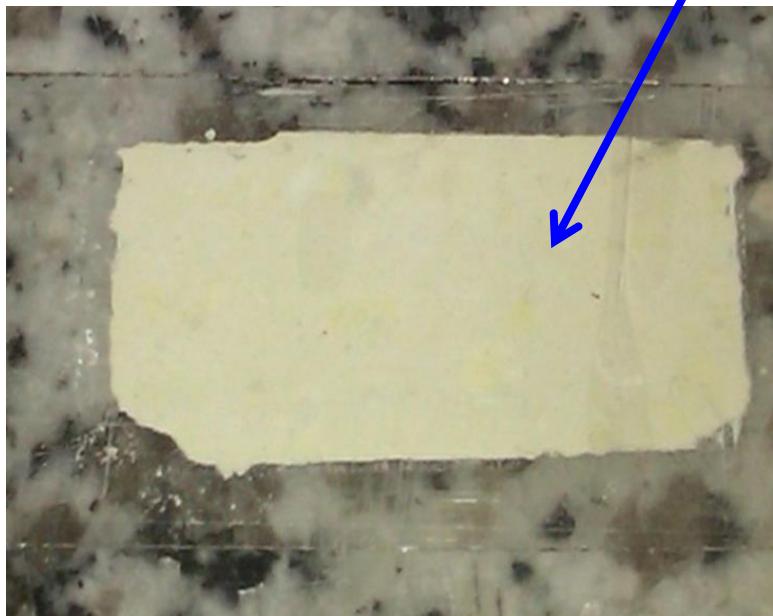


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Synthesis of SrS:Ce

- $\text{Sr}^{2+} + \text{SO}_4^{2-} \rightarrow \text{SrSO}_4 \downarrow$ ($\text{SrSO}_4:\text{Ce}$ 0.05% and 0.5%) \downarrow
- $\text{SrSO}_4:\text{Ce}$ reduced to SrS:Ce (CO atmosphere, 1000 C)



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SrS:Ce Sintering

- Temperatures:

1000°C, 1400°C, 1500°C, 1600°C, 1700°C



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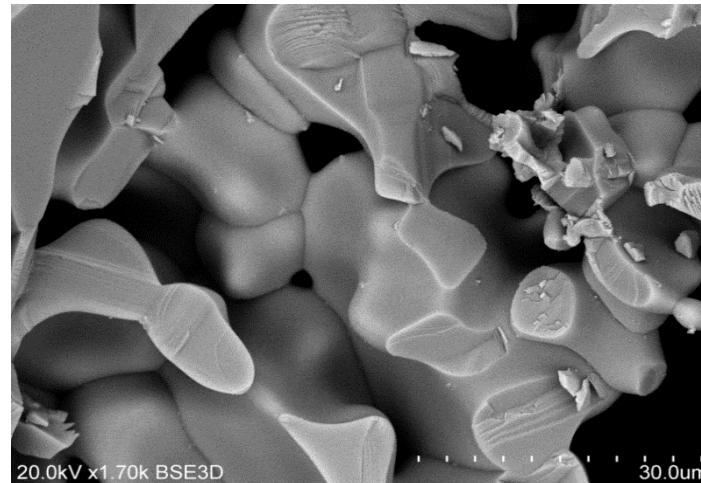
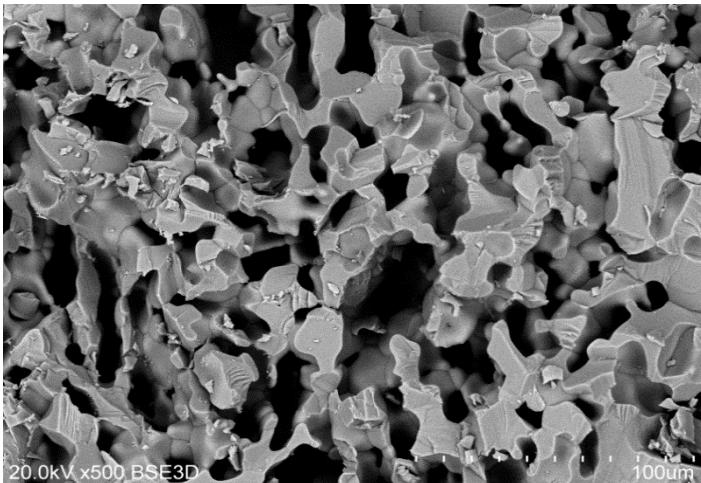
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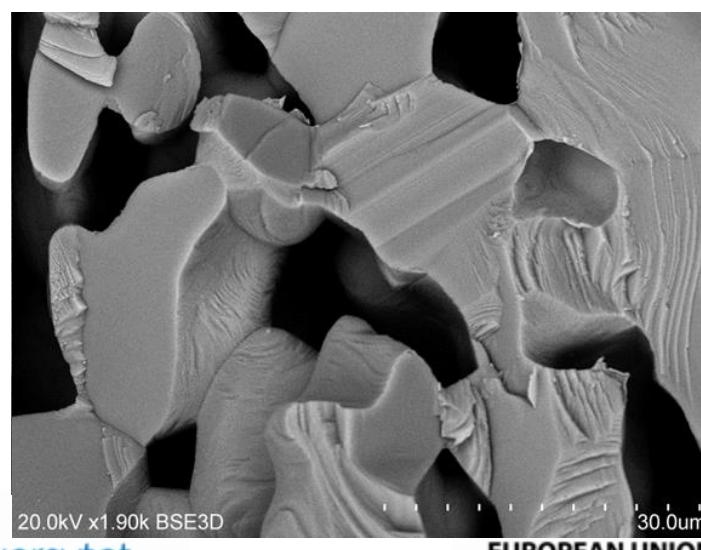
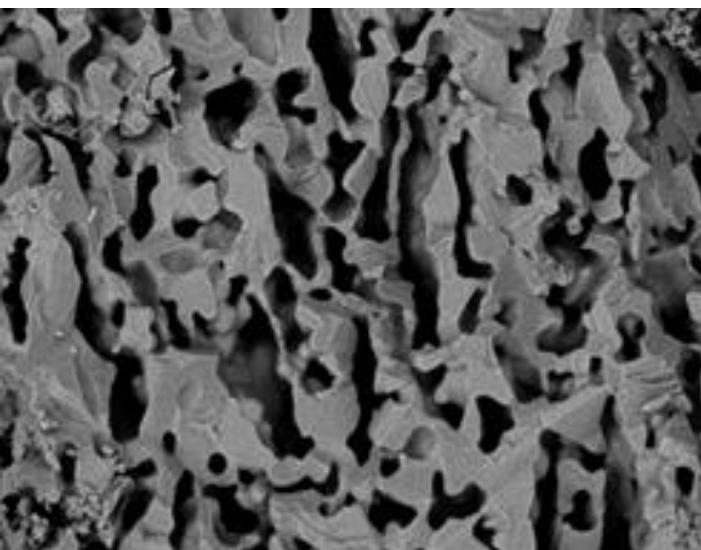


SINTERED CERAMICS - 1400/1500 °C

1400 °C



1500 °C



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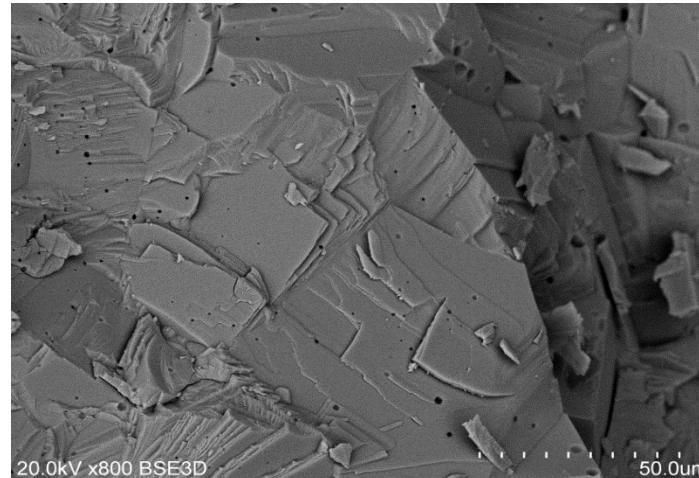
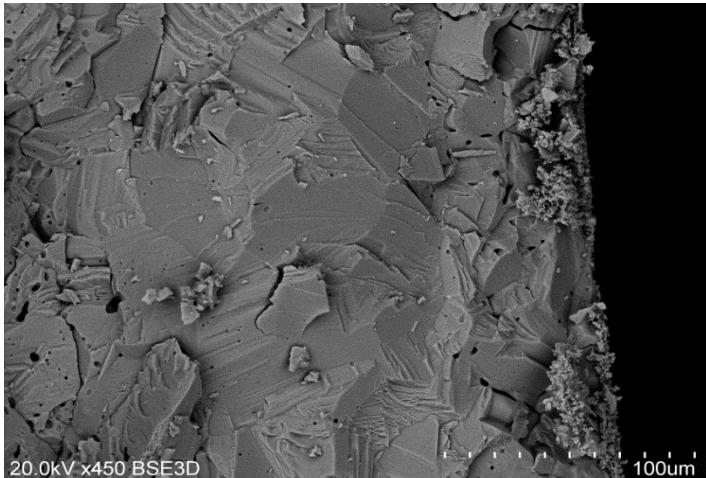
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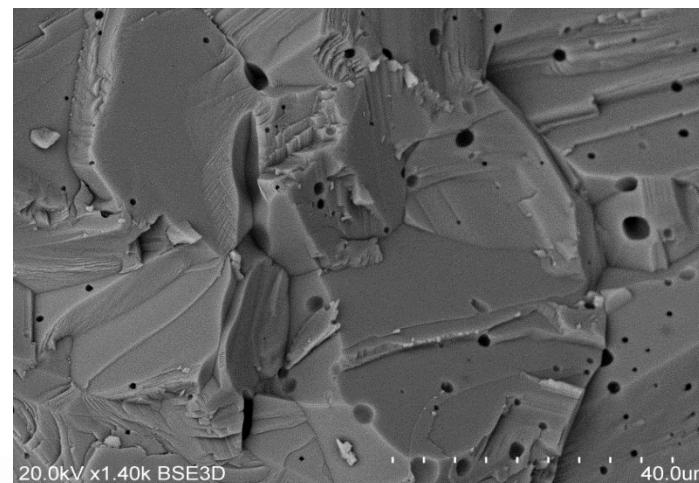
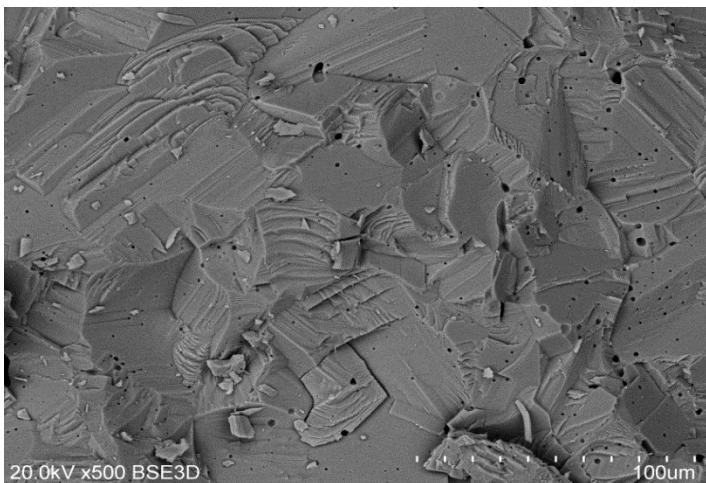


SINTERED CERAMICS - 1600/1700 °C

1600 °C



1700 °C



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Spectroscopic Properties of the Sintered SrS:Ce



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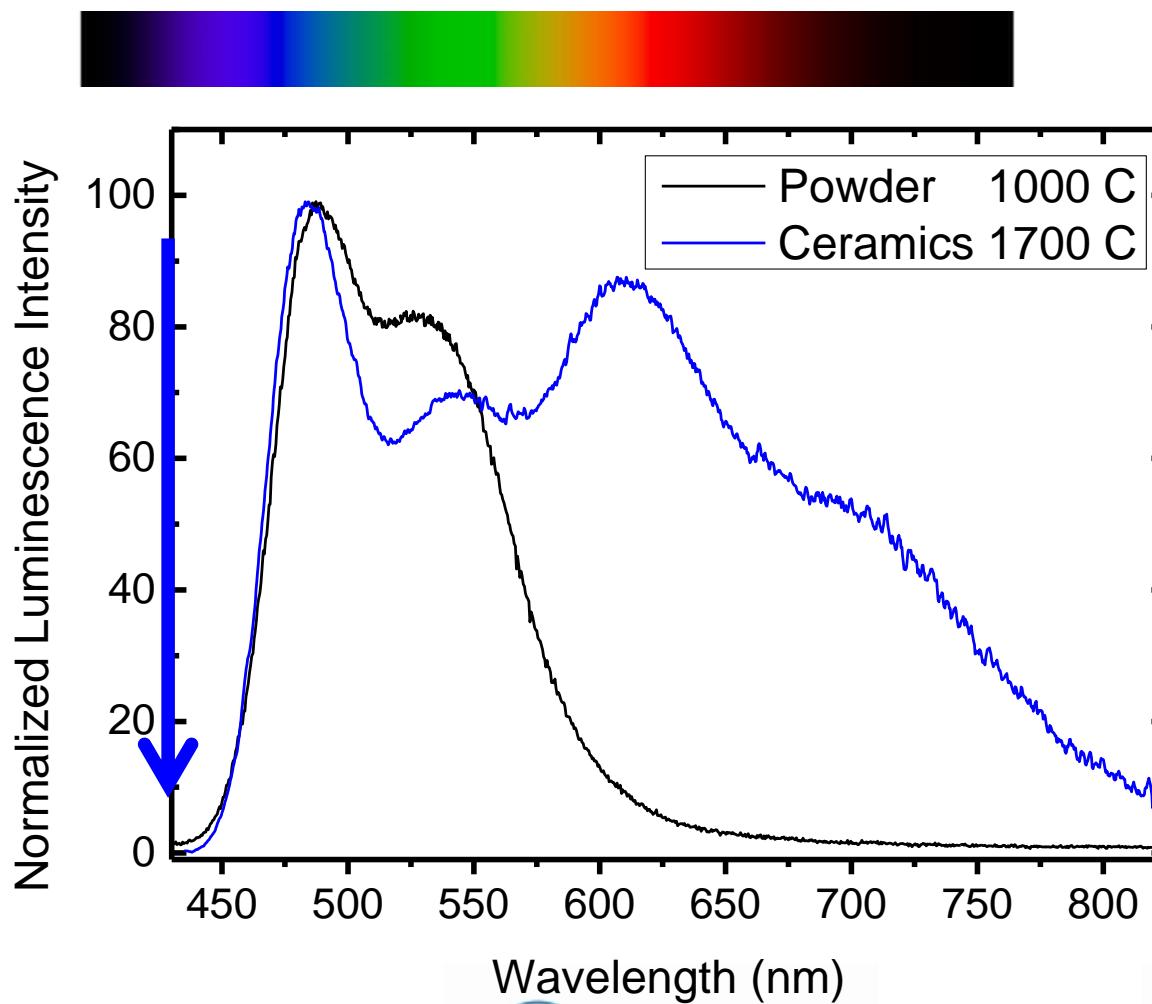


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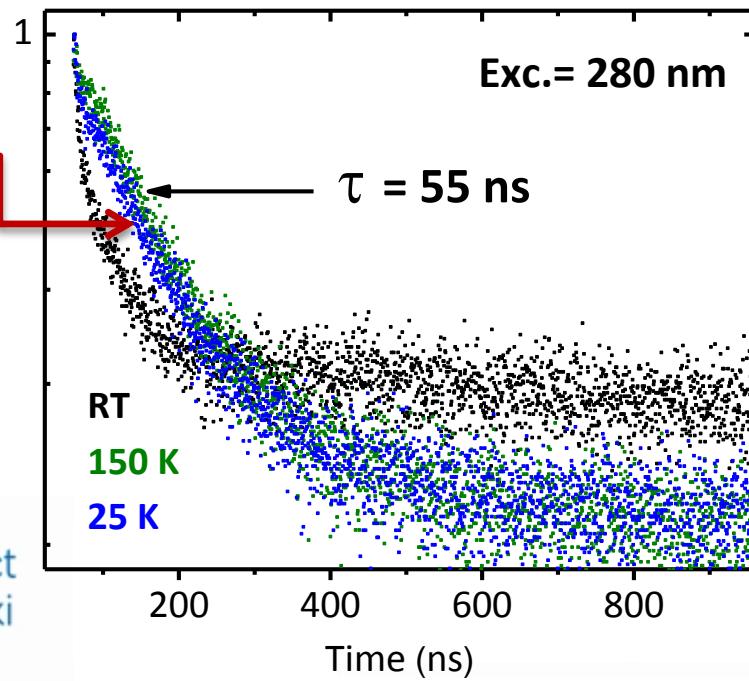
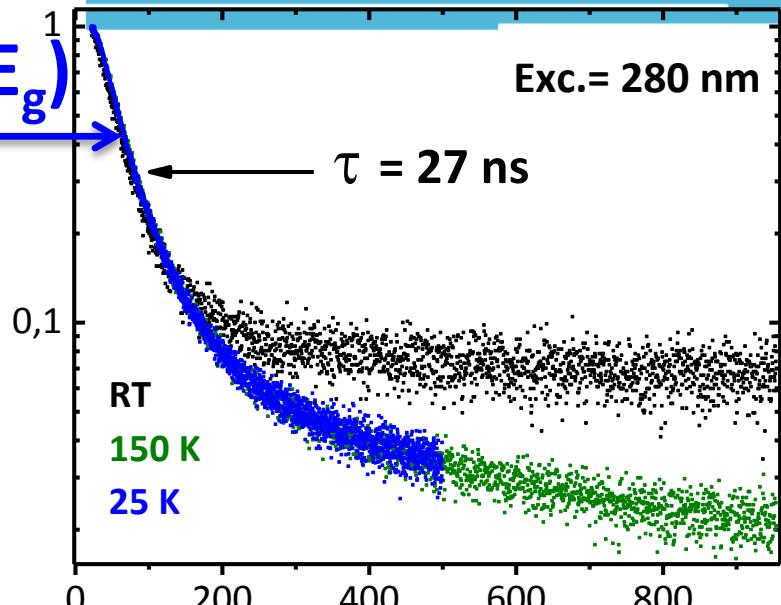
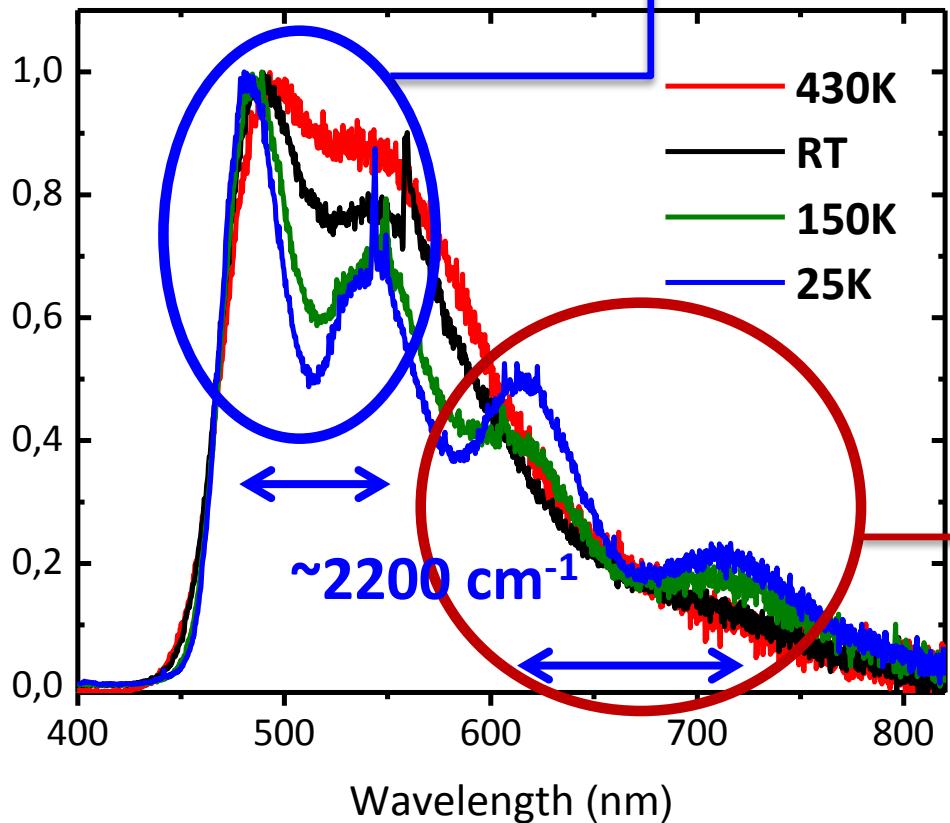
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0.05%Ce, PL - Exc.= 280 nm ($\sim E_g$)

Normalized Intensity

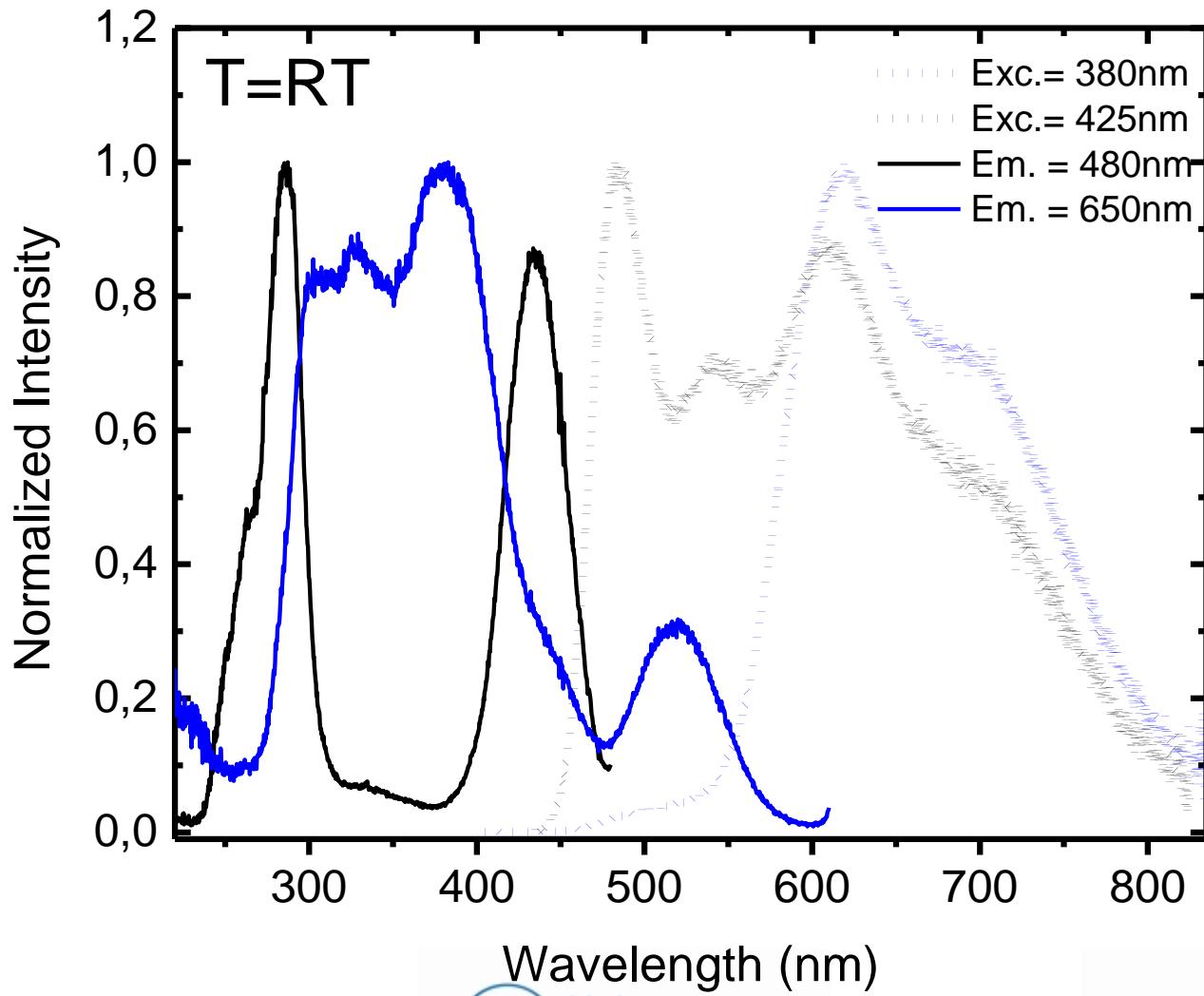


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0.05%Ce, Different Emitting Centers - RT



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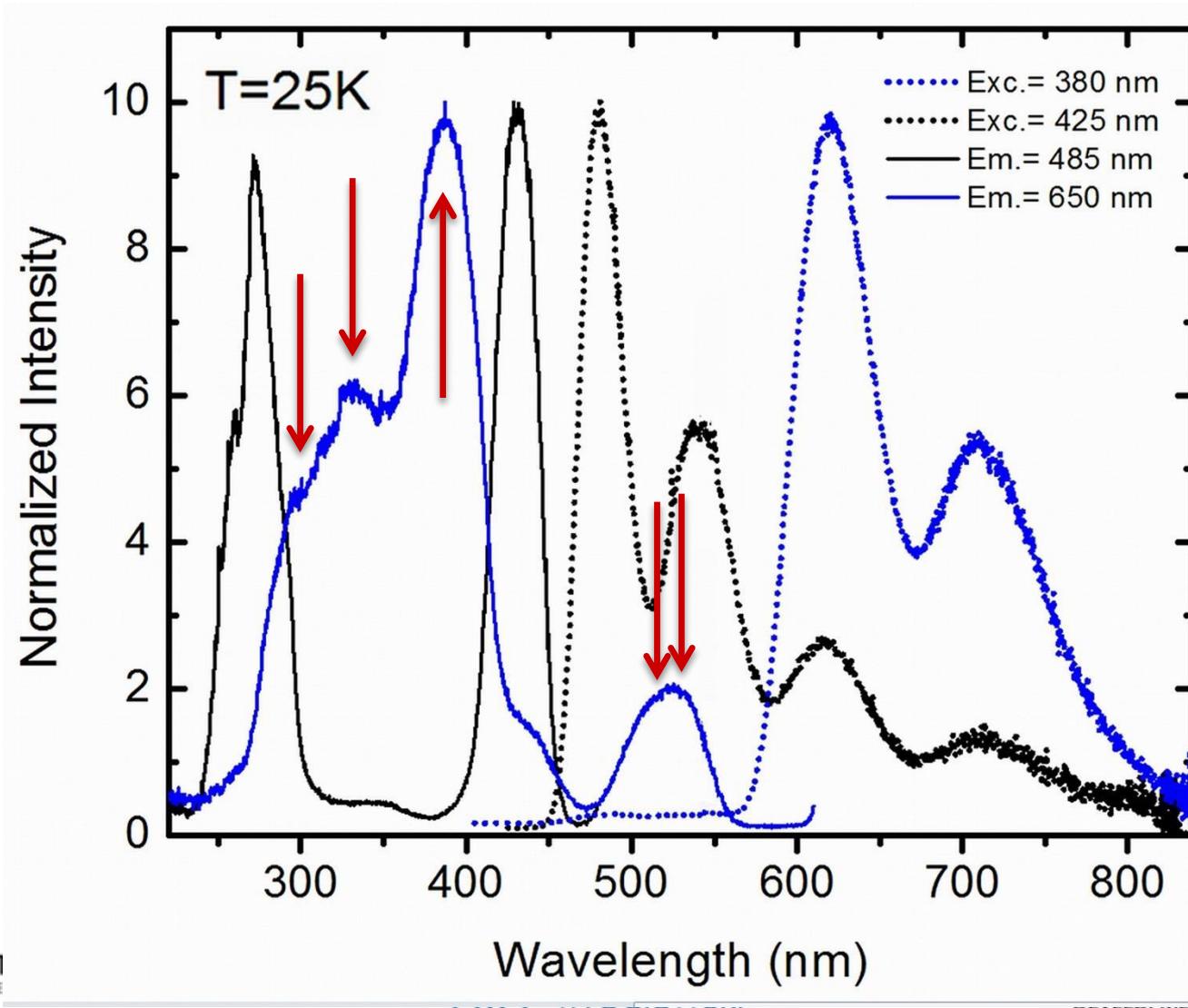


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0.05%Ce, Different Emitting Centers - 25 K



**The new center has low symmetry
as all 5d orbitals of Ce³⁺ have different energies**



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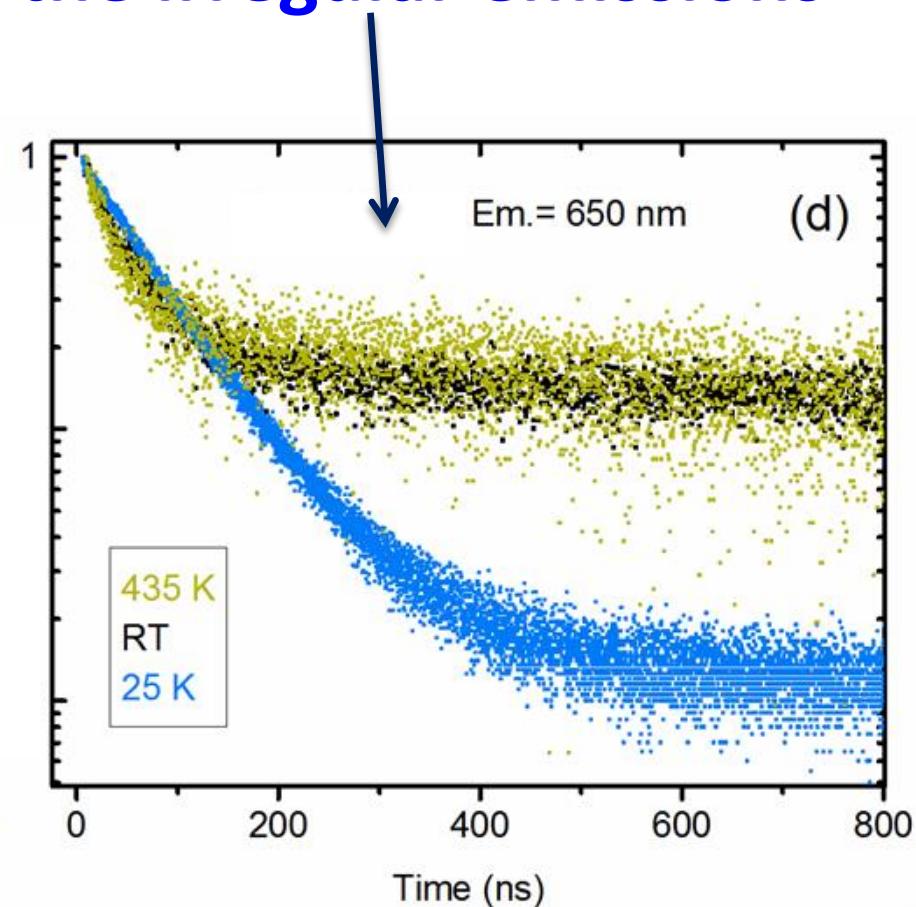
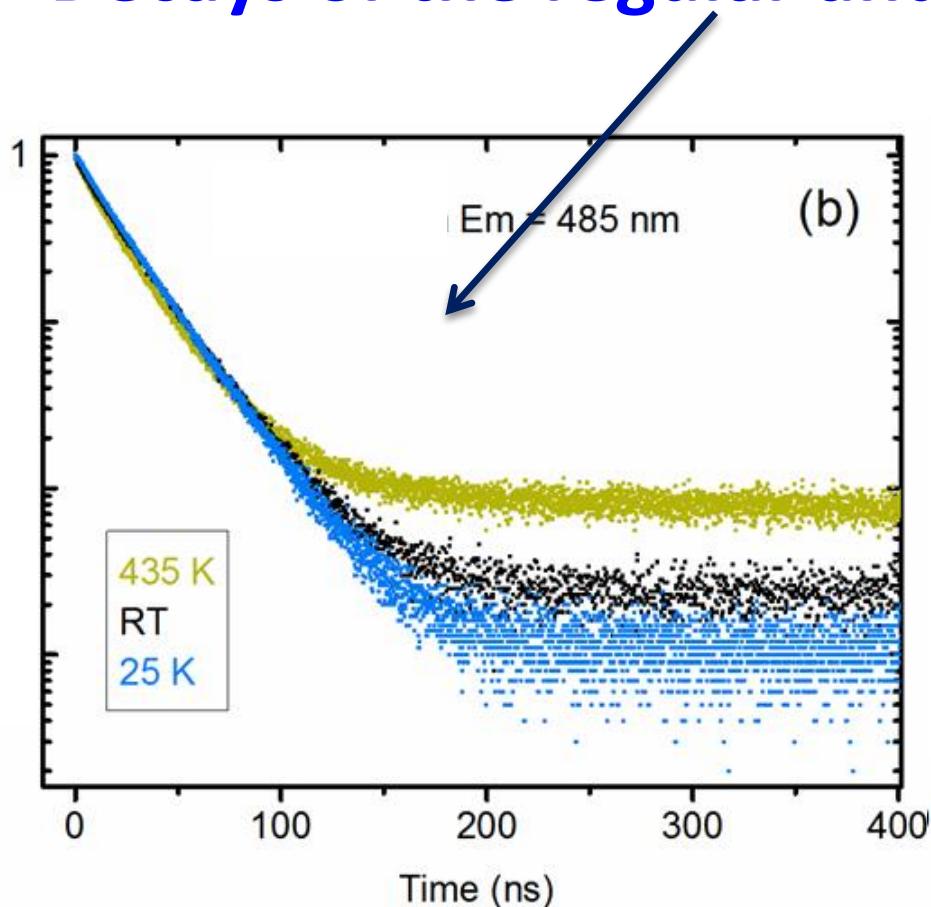
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No energy transfer between the two centers

Decays of the regular and the irregular emissions



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Upon the properties of the new emitting center we propose that it cosist of clusters:



Interstitial S^{2-}

balancing the extra charge
of two Ce^{3+}



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The clustering occurs
only upon high-temperature sintering as then
an efficient mass transfer allows the defect ions to
migrate for distances long enough to form such
complex structures.

Interstitial S²⁻ lowers the local symmetry of Ce³⁺
engaged into formation of the



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Conclusions

- High-temperature sintering (1700 °C) of SrS:Ce leads to a new red-IR Ce³⁺ emission band
- Both centers have much different excitation spectra (site-selective excitation possible)
- The red-emitting Ce³⁺ has much lower local symmetry than the regular one
- There is no energy transfer between the regular and the irregular (red) Ce³⁺ centers



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